



Sebastian Stock^{1,2}, Malina Seyffertitz¹, Nikolaos Kostoglou³, Max Valentin Rauscher¹, Bruno Demé², Viviana Cristiglio², Stephan Rols², Steve Hinder⁴, Mark Baker⁴, Volker Presser⁵, Christian Mitterer³, Oskar Paris¹

¹Chair of Physics, Montanuniversität Leoben; ²Institute Laue Langevin, Grenoble; ³Department of Materials Science, Montanuniversität Leoben; ⁴Department of Mechanical Engineering Sciences, University of Surrey; ⁵Leibniz Institute for New Materials, Department of Materials Science & Engineering, Saarland University

Motivation

Hydrogen will play an important role in the transition from a fossil fuel based economy towards a renewable and CO₂ free one in the future. The hydrogen needed in industrial applications, such as steel making [1], can be stored in large high pressure storage vessels or underground caverns, whereas mobile hydrogen storage solutions for fuel cell electric vehicle require a volume-efficient and light weight structure [2]. The hydrogen storage performance of high surface area activated carbons at cryogenic temperatures offers a possible solution. In order to enhance the performance, an understanding of the basic physical processes of the solid-gas interaction is crucial. Within this work we try to elucidate the effects of pore size and pressure on the density evolution of the confined hydrogen using Small Angle Neutron Scattering and the insights from a hierarchical contrast model.

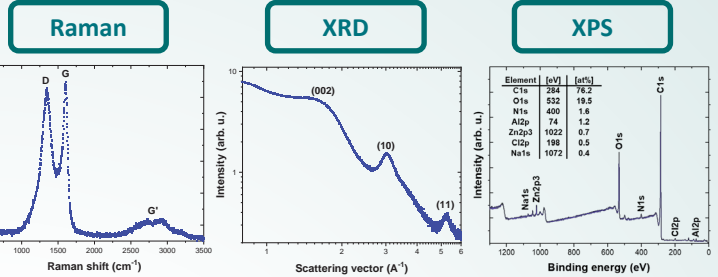
Materials characterization

Thorough materials characterization enables us to understand the structure of the material from a macroscopic scale down to the atomistic scale.

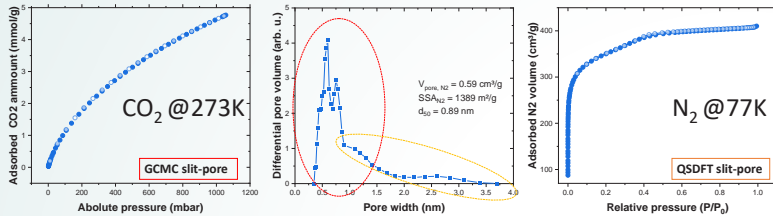
Gas Sorption Analysis is a fast and non-destructive technique, which elucidates the pore structure, the surface area as well as the pore volume [3].

Raman spectroscopy and X-ray diffraction helps to understand the atomistic structure of the material [4].

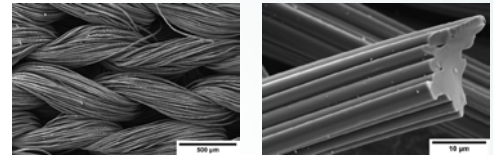
X-ray photoelectron spectroscopy is a powerful tool to assess the chemical composition of the activated carbon [5].



Gas Adsorption

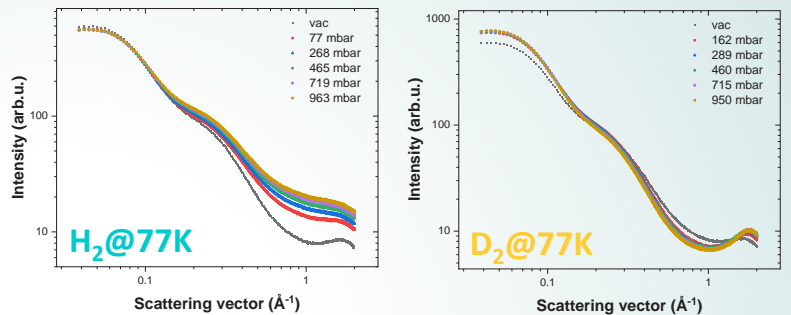


Scanning electron microscopy

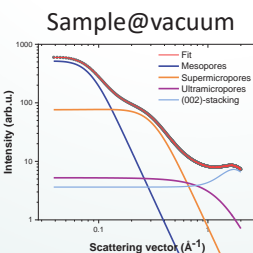
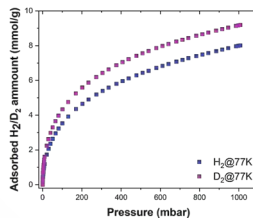


Combining Gas Adsorption and Small Angle Neutron Scattering

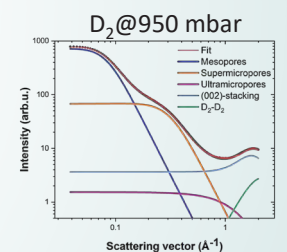
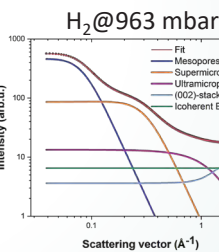
Small momentum transfer diffractometer D16@ILL



The hierarchical pore scattering contrast model



$$I(q)_{\text{pore}} = \frac{S_d(p) \cdot T^2}{\tilde{I} \cdot q^2} \cdot \frac{[(qT)^2 + (a-1) \cdot (a^2 + b^2)]}{[(qT)^2 + a^2 - b^2]^2 + 4a^2b^2}$$



$$\frac{S_d(p)}{S_d(0)} = \frac{\varphi_d \cdot \varphi_{\text{matrix}} \cdot \langle \tilde{\rho}_{\text{matrix}} \rangle - \tilde{\rho}_d^2}{\varphi_d \cdot \varphi_{\text{matrix}} \cdot \langle \tilde{\rho}_c \rangle}$$

$$\rho_{\text{gas}} = \frac{M_{\text{H}_2/\text{D}_2} \cdot \tilde{\rho}_d}{b_{\text{H}_2/\text{D}_2} \cdot N_A} \left(\frac{\text{g}}{\text{cm}^3} \right)$$



References

- [1] D. Ernst, et al., Plasma Smelting Reduction Process, (2023). Metals 2023, 13(3), 558; <https://doi.org/10.3390/met13030558>
- [2] D.P. Broom, D. Book, Hydrogen storage in nanoporous materials, in: Adv. Hydrog. Prod. Storage Distrib., Elsevier Inc., 2014; pp. 410–450. <https://doi.org/10.1533/9780857097736.3.410>.
- [3] M. Thommes, et al., Physisorption of gases, with special reference to the evaluation of surface area and pore size distribution (IUPAC Technical Report), Pure Appl. Chem. 87 (2015) 1051–1069. <https://doi.org/10.1515/pac-2014-1117>.
- [4] S. Stock, et al., Coffee Waste-Derived Nanoporous Carbons for Hydrogen Storage, ACS Appl. Energy Mater. (2022). <https://doi.org/10.1021/acsaem.2c01573>.
- [5] N. Kostoglou, et al., Nanoporous activated carbon cloth as a versatile material for hydrogen adsorption, selective gas separation and electrochemical energy storage, Nano Energy (2017). <https://doi.org/10.1016/j.nanoen.2017.07.056>.

SyNergy_Mat Lab
Chair of Physics

